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for two people. Synthesis and characterization of sodium alginate-chitosan-iron (III) oxide nanocomposite for the reduction of Cr(VI) and their application in the removal of chromium ions from aqueous solution. The synthesis of the novel ferric oxide/alginate/chitosan/Fe(3+) nanocomposite and a detailed investigation of its structural, morphological, spectral and adsorption properties towards the elimination of Cr(VI) in aqueous solution are presented in this work. The nanocomposite was synthesized by a simple co-precipitation method using sodium alginate, chitosan and Fe(3+) followed by cross-linking with calcium chloride. The structural and morphological characterizations of the nanocomposite were studied using different techniques. The synthesized nanocomposite was characterized by Fourier transform infrared spectroscopy (FTIR), X-ray diffraction (XRD), scanning electron microscopy (SEM) and particle size analyzer (PSA). The composite material was investigated to determine the effect of chemical variables such as pH, contact time and initial chromium ion concentration on the removal of chromium ions from aqueous solution. The results showed that by increasing the pH values of the solution both adsorption kinetics and equilibrium were increased. The maximum adsorption capacity was calculated to be about 0.44 mg/g. The results also showed that the adsorption isotherm is of the Langmuir type and can be well described by the Temkin model. The thermodynamic parameters suggest that the adsorption of Cr(VI)